ANSWERS TO "TRY YOURSELF" PROBLEMS FROM STUDY SECTION 8.4

Try Yourself 8.4 a

What is the pK_a value for the conjugate acid of ammonia?

- a) 4.74
- b) 9.25
- c) 5.60
- d) 7.00

Try Yourself 8.4 b

Which of the following has a $\ensuremath{\mathsf{pK}}_a$ value of 4.20?

- a) Benzoic acid, C₆H₅CO₂H
- b) Acetic acid, CH₃CO₂H
- c) Metanoic acid, HCO₂H
- d) Hydrofluoric acid, HF